A New and Efficient Synthesis of Highly Functionalized (Triphenylphosphoranylidene)cyclopentadienes with Zwitterionic Resonance Structures

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Different esters of 2-aryl-4-hydroxy-3-nitro-5-(triphenylphosphoranylidene)cyclopenta-1,3-diene-1-carboxylates **1** were prepared in excellent yields from the 1:1:1 reaction between Ph₃P, dialkyl acetylenedicarboxylates **2**, and (substituted) 2-(nitroethenyl)benzenes **3** (*Scheme 1*). The structures of the highly functionalized compounds **1** were corroborated spectroscopically (IR, ¹H-, ¹³C-, ³¹P-NMR, EI-MS) and by elemental analyses. A plausible mechanism for this type of cyclization is proposed (*Scheme 2*).

Introduction. – Phosphorus ylides are important reagents in organic chemistry, especially in the synthesis of naturally occurring products with biological and pharmacological activities [1]. (Triphenylphosphoranylidene)cyclopentadienes are usually prepared by deprotonation of related phosphonium salts; the latter, in turn, are prepared by the reaction of Ph₃P and a cyclopentadienyl halide [2]. Cyclopentadiene-based ylides are of interest to us, because these compounds may serve as ligands in the synthesis of many inorganic complexes, and since they are usually stable and quite unreactive for ylide species [3].

Zwitterionic species are known to arise from the addition of nucleophiles such as Ph₃P [4], pyridine [5], a wide range of tertiary amines [6], and isocyanide [7] to activated acetylenes [8]. Earlier attempts to trap these species with dipolarophiles to generate cyclic compounds have been essentially unsuccessful [9]. We have recently shown that these species can be efficiently trapped by dipolarophiles such as aryl aldehydes [10], 'isopropylidene Meldrum's acid' (=2,2-dimethyl-5-(1-methylethylidene)-1,3-dioxane-4,6-dione) [11], and aryl isocyanates [12], leading to a facile synthesis of 1,3-diionic organophosphorus compounds, 1,5-diionic organophosphorus compounds, and β -lactam derivatives, respectively. Considering the above background and also the importance of conjugate addition of nucleophiles to electron-deficient alkenes for the creation of carbon-carbon or carbon-heteroatom bonds, we surmised that the zwitterions might undergo cycloaddition to activated alkenes, thus constituting a synthesis of highly substituted (triphenylphosphoranylidene)cyclopentadienes. We now report an efficient synthesis of 2-aryl-4-hydroxy-3-nitro-5-(triphenylphosphoranylidene)cyclopenta-1,3-diene-1-carboxylates from Ph₃P, dialkyl acetylenedicarboxylates **2**, and substituted (2-nitroethenyl)benzenes **3** (*Scheme 1*).

Scheme 1

Results and Discussion. – The reaction of dialkyl acetylenedicarboxylates **2** with vinylogous nitrobenzenes **3** in the presence of Ph_3P , proceeded spontaneously at room temperature in anhydrous Et_2O , and was complete within a few minutes. 1H - and ^{13}C -NMR spectra of the crude precipitate clearly indicated the formation of the carboxylates **1**. No other products could be detected by NMR spectroscopy. The structures of compounds 1a-f were deduced from their elemental analyses, IR, 1H -, ^{13}C -, and ^{31}P -NMR spectra. The mass spectra of these compounds displayed molecularion peaks at appropriate m/z values.

The $^1\text{H-NMR}$ spectrum of 1a exhibited two sharp *singlets* arising from one MeO ($\delta(H)$ 2.84) and one OH group ($\delta(H)$ 10.95). The OH H-atom at $\delta(H)$ 10.95 was strongly deshielded as a result of extensive intramolecular H-bonding. The Ph residues gave rise to characteristic signals in the aromatic region of the spectrum.

The ¹H-decoupled ¹³C-NMR spectrum of **1a** showed five characteristic *doublets* at $\delta(C)$ 68.55 (¹J(C,P) = 121.9 Hz), 11.62 (²J(C,P) = 7.1 Hz), 122.94 (³J(C,P) = 17.2 Hz), 129.02 (²J(C,P) = 8.9 Hz), and 159.18 (³J(C,P) = 8.4 Hz) for C(5), C(1), C(3), C(4), and C(2), respectively. The ¹³C NMR spectrum was in agreement with the anticipated structure **1**. Partial assignments of these resonances are given in the *Exper. Part.* The proton-decoupled ³¹P-NMR spectrum of **1a** exhibited sharp signals at $\delta(P)$ 12.64. The ¹H- and ¹³C-NMR spectra of the **1b**-**f** were similar to those of **1a**, except for the ester groups, which exhibit characteristic signals with appropriate chemical shifts (see *Exper. Part*).

Although the mechanism of the reaction between 2 and 3 in the presence of Ph₃P has not yet been established experimentally, a possible explanation is proposed in *Scheme 2*. On the basis of the well-established chemistry of trivalent phosphorus nucleophiles [13–20], it is reasonable to assume initial addition of Ph₃P to dialkyl acetylenedicarboxylate, followed by attack of the resulting zwitterion 4 on the *Michael* acceptor 3 to yield betaine 5, which then cyclizes. Finally, the resulting cation 6 loses ROH to produce compound 1.

In conclusion, the present method has the advantage that the reaction can be performed under neutral conditions, and that the substances can be mixed without any activation or modification. The simplicity of this procedure makes it an interesting alternative to complex, multistep approaches.

Experimental Part

General. All reagents and solvents were obtained from Fluka (Buchs, Switzerland) and used without further purification. Compounds $3\mathbf{a} - \mathbf{c}$ were prepared according to a published procedure [21]. Melting points (m.p.): Electrothermal 9100 apparatus; uncorrected. IR spectra: Shimadzu IR-460 spectrometer; in cm⁻¹. ¹H-, ¹³C-, and ³¹P-NMR Spectra: recorded at 500.1, 125.7, and 202.4 MHz, resp., on a Bruker DRX-500-AVANCE instrument; in CDCl₃; δ in ppm rel. to SiMe₄ (=0 ppm), J in Hz. EI-MS (70 eV): Finnigan MAT-8430 mass spectrometer; in m/z (rel. %). Elemental analyses: Heraeus CHN-O-Rapid analyzer.

General Procedure for the Preparation of Compounds 1. To a magnetically stirred soln. of 2 (1 mmol) and 3 (1 mmol) in anh. Et_2O (5 ml) was added dropwise a soln. of Ph_3P (0.26 g, 1 mmol) in Et_2O (3 ml) at r.t. over 10 min. The mixture was stirred for 1 h, the resulting solid was filtered off, washed with anh. Et_2O , and dried in vacuo to afford 95-98% of 1.

Methyl 4-Hydroxy-3-nitro-2-phenyl-5-(triphenylphosphoranylidene)cyclopenta-1,3-diene-1-carboxylate (1a). Yield: 98%. Orange powder. M.p. $130-132^{\circ}$ (dec.). IR (KBr): 3410 (OH), 1698 (CO2 Me), 1655 (C=C), 1544 (Ph), 1415 and 1374 (NO₂), 1290 (C−O of C−OH), 1201 (C−O of ester), 1105 and 991 (P−Ph). ¹H-NMR (500.1 MHz, CDCl₃): 2.84 (s, 3 H); 7.28 (dd, J=6.8, 6.8, 1 H); 7.33 −7.38 (m, 4 H); 7.54 −7.56 (m, 6 H); 7.65 (dd, J=7.2, 7.2, 3 H); 7.73 (dd, J=13.0, 7.7, 6 H); 10.95 (s, 1 H). ¹³C-NMR (125.7 MHz, CDCl₃): 49.41; 68.55; 111.62; 122.94; 124.36; 125.89; 126.43; 128.30; 129.02; 129.26; 132.00; 133.19; 133.65; 159.18; 164.71. ³¹P-NMR (202.4 MHz, CDCl₃): 12.64 (Ph₃P=C). EI-MS: 521 (1, M^+), 473 (1), 432 (1), 403 (1), 386 (1), 363 (2), 329 (1), 304 (2), 278 (59), 262 (100), 183 (73), 152 (14), 108 (28), 77 (30), 51 (14). Anal. calc. for C₃¹H₂₄NO₃P (521.50): C 71.40, H 4.64, N 2.69; found: C 71.3, H 4.5, N 2.6.

Ethyl 4-Hydroxy-3-nitro-2-phenyl-5-(triphenylphosphoranylidene)cyclopenta-1,3-diene-1-carboxylate (**1b**). Yield: 95%. Yellow powder. M.p. $116-118^{\circ}$ (dec.). IR (KBr): 3425 (OH), 1708 (CO₂Et), 1630 (C=C), 1528 (Ph), 1429 and 1380 (NO₂), 1263 (C=O of C=OH), 1233 (C=O of ester), 1097 and 995 (P=Ph). ¹H-NMR (500.1 MHz, CDCl₃): 0.48 (t, J = 6.8, 3 H); 3.35 (q, J = 6.8, 2 H); 7.30 (dd, J = 7.2, 7.2, 1 H); 7.35 (d, J = 7.0, 2 H); 7.45 (d, J = 6.5, 2 H); 7.52 (m, 6 H); 7.62 (dd, J = 6.7, 6.7, 3 H); 7.71 (dd, J = 13.0, 7.8, 6 H); 10.94 (g, 1 H). ¹³C-NMR (125.7 M, CDCl₃): 12.64; 58.70; 68.65; 112.25; 122.95; 124.43; 125.74; 126.36; 128.30; 128.65; 129.32; 132.30; 133.26; 136.03; 159.14; 164.66. ³¹P-NMR (202.4 MHz, CDCl₃): 12.64 (Ph₃P=C). EI-MS: 535 (1, M⁺), 447 (1), 369 (1), 320 (1), 262 (100), 245 (17), 217 (10), 199 (10), 183 (79), 152 (17), 125 (107), 108 (79), 77 (17), 47 (68), 31 (100). Anal. calc. for $C_{32}H_{26}NO_{3}P$ (535.48): C 71.77, H 4.89, N 2.61; found: C 71.6, H 4.8, N 2.6.

Methyl 4-Hydroxy-3-nitro-2-(4-nitrophenyl)-5-(triphenylphosphoranylidene)cyclopenta-1,3-diene-1-carboxylate (**1c**). Yield: 98%. Yellow powder. M.p. 185 −187° (dec.). IR (KBr): 3400 (OH), 1695 (CO₂Me), 1687 (C=C), 1541 (Ph), 1504 and 1382 (NO₂), 1292 (C−O of C−OH), 1203 (C−O of ester), 1116 and 963 (P−Ph). 1 H-NMR (500.1 MHz, CDCl₃): 2.83 (5 8, 3 H); 7.52 (5 8, 2 H); 7.57 (5 8, 6 H); 7.68 (5 8, 6 Hz, 6 Hz, 7 Hz, 6 Hz, 7 Hz, 6 Hz, 7 Hz, 6 Hz, 7 Hz,

Ethyl 4-Hydroxy-3-nitro-2-(4-nitrophenyl)-5-(triphenylphosphoranylidene)cyclopenta-1,3-diene-1-carboxylate (1d). Yield: 95%. Yellow powder. M.p. $184-185^{\circ}$ (dec.). IR (KBr): 3400 (OH), 1682 (CO2 Me), 1587 (C=C), 1543 (Ph), 1502 and 1342 (NO₂), 1292 (C=O of C=OH), 1210 (C=O of ester), 1109 and 993 (P=Ph). 1 H-NMR (500.1 MHz, CDCl₃): 0.47 (t, J = 6.8, 3 H); 3.36 (q, J = 6.8, 2 H); 7.52 (d, J = 7.7, 2 H); 7.56 (m, 6 H); 7.67 (dd, J = 6.8, 6.8, 3 H); 7.73 (dd, J = 12.8, 7.7, 6 H); 8.19 (d, J = 7.7, 2 H); 10.70 (s, 1 H). 13 C-NMR (125.7 MHz, CDCl₃): 12.64; 59.00; 69.80; 112.47; 121.65; 122.64; 123.99; 125.95; 128.30; 130.50; 132.46; 133.20; 143.90; 145.83; 158.91; 164.01. 31 P-NMR (202.4 MHz, CDCl₃): 12.88 (Ph₃P=C). EI-MS: 580 (1, M+), 537 (1), 522 (1), 479 (2), 463 (2), 445 (1), 369 (1), 325 (1), 304 (1), 277 (100), 262 (86), 199 (20), 183 (63), 108 (16), 77 (16), 51 (16). Anal. calc. for C_{32} H₂₅N₂O₇P (580.53): C 66.21, H 4.34, N 4.83; found: C 66.1, H 4.3, N 4.7.

Methyl 2-(4-Chlorophenyl)-4-hydroxy-3-nitro-5-(triphenylphosphoranylidene)cyclopenta-1,3-diene-1-carboxylate (**1e**). Yield: 98%. Orange powder. M.p. 182−183° (dec.). IR (KBr): 3420 (OH), 1702 (CO2 Me), 1676 (C=C), 1549 (Ph), 1467 and 1373 (NO₂), 1285 (C−O of C-OH), 1209 (C−O of ester), 1100 and 963 (P−Ph). 1 H-NMR (500.1 MHz, CDCl₃): 2.86 (s, 3 H); 7.3 (s, 4 H); 7.55 (m, 6 H); 7.66 (dd, J = 6.7, 3 H); 7.72 (dd, J = 13.1, 7.5, 6 H); 10.86 (s, 1 H). 13 C-NMR (125.7 MHz, CDCl₃): 49.45; 69.93; 111.77; 122.82; 124.19; 126.64; 127.44; 128.28; 130.79; 131.71; 132.30; 133.14; 134.16; 159.10; 164.52. 31 P-NMR (202.4 MHz, CDCl₃): 12.72 (Ph₃P=C). EI-MS: 555 (1, M⁺), 510 (1), 479 (1), 461 (1), 445 (1), 423 (2), 370 (1), 324 (2), 278 (100), 262 (100), 199 (23), 183 (100), 152 (20), 108 (34), 77 (10). Anal. calc. for C₃1H₂3CINO₃P (555.95): C 66.97, H 4.17, N 2.52; found: C 66.8, H 4.1, N 2.5.

Ethyl 2-(4-Chlorophenyl)-4-hydroxy-3-nitro-5-(triphenylphosphoranylidene)cyclopenta-1,3-diene-1-carboxylate (1f). Yield: 95%. Yellow powder. M.p. 174–175° (dec.). IR (KBr): 3400 (OH), 1682 (CO₂Et), 1676 (C=C), 1547 (Ph), 1465 and 1379 (NO₂), 1291 (C-O of C-OH), 1207 (C-O of ester), 1107 and 989 (P-Ph). 1 H-NMR (500.1 MHz, CDCl₃): 0.53 (t, J = 7.0, 3 H); 3.39 (q, J = 7.0, 2 H); 7.31 (s, 4 H); 7.56 (m, 6 H); 7.65 (dd, J = 6.6, 6.6, 3 H); 7.73 (dd, J = 13.0, 7.8, 6 H); 10.88 (s, 1 H). 13 C-NMR (125.7 MHz, CDCl₃): 13.16; 59.30; 69.40; 112.77; 123.88; 124.76; 126.99; 127.52; 128.67; 131.31; 132.01; 132.73; 133.64; 135.04; 159.50; 164.85. 31 P-NMR (202.4 MHz, CDCl₃): 12.51 (Ph₃P=C). EI-MS: 569 (1, M^+), 525 (2), 468 (1), 423 (1), 397 (3), 379 (1), 360 (1), 310 (2), 278 (100), 262 (56), 199 (13), 183 (76), 152 (20), 108 (56), 77 (33), 44 (100). Anal. calcd. for $C_{32}H_{25}$ CINO₅P (569.97): C 67.43, H 4.42, N 2.46; found: C 67.3, H 4.3, N 2.4.

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